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9.1 KMT_POGIL_solutions 17.pdf - Kinetic Molecular Theory ...

Kinetic Molecular Theory ©POGIL – 2005 5/5 Authored by Applications 1. There is a government warning on all aerosol cans that states: Do not store at a temperature above 120° F (50°C). a) Explain why this warning is required in terms of the relationship between temperature and pressure and the kinetic molecular theory.

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Kinetic Molecular Theory Kinetic Molecular Theory (KMT) allows us to explain the observed macroscopic (measurable) properties of ideal gases (pressure, volume, temperature, etc.) in terms of the microscopic properties of individual gas molecules (or atoms, for monatomic gases). As we will see later, real gases do not follow KMT under all conditions, leading to deviations from the ideal gas law.

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5. What is the relationship between temperature and molecular motion? 6. In terms of the kinetic-molecular theory of gases, how can increase in the temperature of a gas confined in a rigid container cause an increase in the pressure of the gas? spud Will 101/ Intl.) ctn/CZ/h.Lr ofkept- pres Sur= ©POGIL - 2005 4/4 Authored by

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Name_____ Kinetic Molecular Theory Why? The kinetic-molecular theory is a model or a mental image of how particles of matter behave. Knowledge of the kinetic-molecular theory allows us to predict the action of solids, liquids and gases and understand how the changes of state occur. Learning Objectives: Identify the basic differences between particle behavior in the solid, liquid and gaseous ...

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Pogil Naming Molecular Compounds

The kinetic molecular theory is a simple but very effective model that effectively explains ideal gas behavior. The theory assumes that gases consist of widely separated molecules of negligible volume that are in constant motion, colliding elastically with one another and the walls of their container with average velocities determined by their absolute temperatures.

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Kinetic Molecular Theory

Kinetic Molecular Theory Applications 1. There is a government warning on all aerosol cans that states: Do not store at a temperature above 1200 F (5000. a) Explain why this warning is required in terms of the relationship between temperature and pressure and the kinetic molecular theory. hljhhq -flu (enja//l.u- more QUjctI prrjw<-

Pogil Kinetic Molecular Theory Answer Key

POGIL: Kinetic Molecular Theory Learning Objectives Identify the basic differences between particle behavior in the solid, liquid, and gaseous phases. Develop an understanding of the postulates of the kinetic molecular theory. Identify how temperature affects molecular motion.

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