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REACTIONS Law of

Conservation of Matter:

Humanly speaking,

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matter is neither created  
nor  
destroyed...therefore in  
a chemical reaction,  
both sides (reactants  
and products) must  
contain the same  
number of atoms of each  
element.

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in this set (48) chemical  
equation. a symbolic  
way of describing a  
chemical reaction.  
skeleton equation. a  
chemical equation that  
does not indicate the  
amounts of substances

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involved. catalyst. a  
substance that speeds ...

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Answer the following in  
the space provided. 22.  
Identify the type of each  
of the following  
reactions. a.  $2\text{C} + 6\text{H} \dots$



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SECTION 11.1

DESCRIBING

CHEMICAL

REACTIONS 1. Write the skeleton equation for the reaction between hydrogen and oxygen that

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Section 11.1 –

Describing Chemical Reactions In a chemical reaction, the reactants are written on the left and the products on the right. The arrow that separates them is called yield.

Chapter 11

Although the chemical reactions involved in

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photosynthesis and digestion are different, both chemical reactions are necessary to sustain life. In a chemical reaction, one or more reactants change into one or more products. Figure 11.1a shows the ingredients for making leavened bread— flour, salt, yeast, and water.

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Humanlyspeaking,

destroyed.,bothsides ...

11.2 Types of Chemical

Reactions> 29 Whether

one metal will displace

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another metal from a compound depends upon the relative reactivities of the

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charcoal briquettes ...  
Chemical Reactions 331

Answers to... Figure

11.5 The reaction  
produces bright, white  
light. Gifted and  
Talented Combination  
reactions are sometimes  
referred to as synthesis  
reactions, but synthe-

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