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Chapter 12 Stoichiometry 12.1 The Arithmetic of Equations 12.2  
Chemical Calculations 12.3 Limiting Reagent and ... Sample Problem  
12.1 When using conversion factors, remember to ... excess of 1800 is  
a logical answer. • The unit of the known (FSW 3 HP 2) cancels.

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of Equations ... Stoichiometry 379 CHAPTER 12 Assessment 36. a. Two  
formula units  $KClO_3$  decom- pose to form two formula units  $KCl$  and three  
molecules  $O_2$ . b. ... Acceptable answers include the idea

Chapter 12.2: Stoichiometry of Reactions in Solution ...  
Chapter 12 Stoichiometry . In the reaction represented by the equation  
 $2Na_2H_2O -y 2NaOH + H_2$ , how many grams of ... Answer the questions  
above, assuming we started with 30 grams of ammonium nitrate and 50  
grams of sodium phosphate. Consider the following reaction:

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 $\text{Mg} + 2\text{H}_2\text{O} \rightarrow \text{Mg}(\text{OH})_2 + \text{H}_2$  ...

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Chapter 11 Small-Scale Lab Section 11.3 Precipitation Reactions: Formation of Solids, page 345 Analysis 1. a.  $\text{Na}_2\text{CO}_3 + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3$

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3 3Ag 2CO 3(s) b. 2Na 3PO 4 3Pb(NO 3) 2 ? 6NaNO 3 Pb 3(PO 4) 2(s) 2.

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Chapter 12 Stoichiometry 127 SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353–358) This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

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