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CHAPTER . 12 . REVIEW ""--

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Solutions . SHORT ANSWER

Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. James made several errors.

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CHAPTER 12 & 13: SOLUTIONS
The following pages contain the bulk (but not all) of the information for the chapter 12 & 13 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document

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and NOTHING else, you should at least be able to PASS the test.

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Solutions ...**

CHAPTER 12 REVIEW Solutions
SECTION 2 SHORT ANSWER

Answer the following questions in the space provided. 1. The following are statements about the dissolving process. Explain each one at the molecular level.

a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas

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12- Solutions**

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University. Modern Chemistry
Chapter 12 Solutions Section 1-
Types of Mixtures Solutions are
homogeneous mixtures

CHAPTER 12 REVIEW

Solutions - Weebly

Modern Chemistry Chapter
12-Solutions Section 1- Types of
Mixtures Solutions are
homogeneous mixtures of two or
more substances in a single
phase. Soluble describes a
substance as capable of being
dissolved. Solvent is the
dissolving medium in a solution.
Solute is the substance that is
dissolved in a solution.

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CHEM 142 REVIEW GUIDE-
CHAPTER 12: SOLUTIONS
Sections 12.1 - 12.4 (Tro, 2 nd
edition) 1. Terminologies:
solution, solute, solvent, miscible
(vs. immiscible) liquids,
dissolution, 2. Know the common

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types of solutions and give examples of each (Table 12.1) 3. Concepts: Be able to explain the following: (a) Intermolecular forces (IMFs) of attractions that exist in solution and relative

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12-Solutions Section 1- Types of
Mixtures Solutions are
homogeneous mixtures of two or
more substances in a single
phase. Soluble describes a
substance as capable of being
dissolved. Solvent is the
dissolving medium in a solution.
Solute is the substance that is
dissolved in a solution.

**Chapter 12 and 13
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Chapter 12 and 13 Review
Worksheet—ANSWERS OGT

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Section Page Person, Place, Date,
Term Description 12.3-13.3 424,
449 Charles Lindbergh Made the
first nonstop solo flight across the
Atlantic Ocean in his plane the
Spirit of St. Louis. He became a
hero upon his return home,
receiving a ticker tap parade and
a reception from the white house.

Modern Chemistry Chapter 12 Review Answer Key

DATE: 12.06.2012 author:
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answers Modern Chemistry
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Chemistry. CHAPTER 1 REVIEW.
Matter and Change. SECTION 1.
SHORT ANSWER Answer the
12. 18. B. 18. 12. 6. A. MODERN

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AND CALCULATIONS. holt
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Modern Chemistry 1 Solutions
CHAPTER 12 REVIEW Solutions
Teacher Notes and Answers
Chapter 12 SECTION 1 SHORT
ANSWER 1. c 2. a 3. b 2. a.
alcohol b. water c. the gels 3. The
mixture is a colloid. The
properties are consistent with
those reported in Table 3 on page
404 of the text. The particle size
is small, but not too small, and
the mixture

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Modern Chemistry Chapter 12-Solutions Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

**CHEM 142 REVIEW GUIDE-
CHAPTER 12: SOLUTIONS**

Modern Chemistry 109 Chapter Test Chapter: Solutions PART I In the space provided, write the letter of the term or phrase that best ... Modern Chemistry 110 Chapter Test Name Class Date Chapter Test B, continued _____ 7. ... 12. 760 mm 13. V P VP 14.

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effusion 15. pressure 16. newton
17. barometer 18. partial
pressure

**Modern Chemistry Chapter 12
Solutions Review Answers**

Modern Chemistry 107 Solutions
CHAPTER 12 REVIEW Solutions
SECTION 3 SHORT ANSWER

Answer the following questions in
the space provided. 1. Describe
the errors made by the following
students in making molar
solutions. a. James needs a 0.600
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out 0.600 g of KCl and adds 1 L of
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section 2 review answers ...**

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