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**Welcome to AP
Chemistry. Chapter
1: Chemical
Foundations.**

**Accuracy and Sig
Figs ... Chapter 17:
Electrochemistry.**

17.1 Galvanic Cells.

**17.2 Standard
Reduction**

**Potentials. 17.3 Cell
Potential, Electrical
Work, and Free
Energy.**

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Chapter 17. STUDY.
PLAY. common-ion
effect. if there is a
weak electrolyte in
equilibrium and you
add an ion that is in***

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common with one of the products in equilibrium, then the weak electrolyte ionizes less than normal (within ICE tables there may be an initial concentration for one or both products)

***AP Chemistry:
Chapter 17 Practice***

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***Test (Key at End)
AP Chemistry -
Agenda for
February/March
2016 Chapter 16
Acid-Base Equilibria
& Chapter 17
Additional Aspects
of Aqueous
Equilibria Ch. 16
Packet Acid-Base
Equilibria***

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and more with
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***Chapter 17 -
Electrochemistry |
CourseNotes***

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Chemistry: The Central Science
Chapter 17:
Additional Aspects of Aqueous Equilibria
Water is important because of its exceptional ability to dissolve wide variety of substances
Two additional types of aqueous equilibria o
Those involving

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***slightly soluble salts
in solutions o Those
involving the
formation of metal
complexes in
solutions***

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Hoggard High***

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***In this video I'll
show you how to
calculate the
hydronium ion
concentration and
final pH in a
common ion effect
problem.***

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**Chapter 17 –
Additional Aspects
of Aqueous
Equilibria: Part 1 of
21
AP Chemistry
Review Questions -
Electrochemistry. ...
10 17 ? 10 2 ? 10-34
? 10 11; This picture
of an
electrochemical cell
can best be**

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***described as: ? a
complete
electrolytic cell ? an
electrolytic cell, but
missing at least one
essential
component ? a
complete galvanic
cell ...***

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AP Chemistry Study
Guide – Chapter 18,
Page 16/33***

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Electrochemistry. Students should be able to... Draw and label the components of a galvanic cell. Be able to write half reactions. Classify reactions as oxidations and reductions, and identify oxidizing and reducing agents. Find the cell

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potential. Build a galvanic cell and measure its potential ...

***AP Chemistry
Review Questions -
Electrochemistry
AP Chemistry
Scoring Guidelines
2017 Author: The
College Board
Subject: AP
Chemistry Scoring***

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Guidelines 2017

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scoring information;

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PM

17.1 Galvanic Cells -

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***Aqueous Equilibria
and the pH equals***

***Plan: (b) We
proceed in the same
way as we did in
part (a), except we
are now past the
equivalence point
and have more OH –
in the solution than
H +.As before, the
initial number of
moles of each
reactant is***

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***determined from
their volumes and
concentrations.***

***Chapter 17
Additional Aspects
of Aqueous
Equilibria
AP Chemistry:
Chapter 17 Practice
Test (Key at End) All
work must be done
on separate paper or
no credit will be***

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***earned. Some work
or explanation must
be shown for EACH
problem or no credit
will be earned for
that question.***

***Ch 17
Thermochemistry
Practice Test
Advanced
Placement
Chemistry: 1996
Free Response***

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Questions 7) $\text{Sr}(s) + \text{Mg}^{2+} \rightleftharpoons \text{Sr} + \text{Mg}(s)$ Consider the reaction represented above that occurs at 25°C . All reactants and products are in their standard states. The value of the equilibrium constant, K_{eq} , for the reaction is 4.2×10^{17} at 25°C .

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**Chemistry: The
Central Science**

**Chapter 17:
Additional ...**

Ch 17

Thermochemistry

Practice Test

Matching Match

**each item with the
correct statement**

below. a. calorimeter

d. enthalpy b.

calorie e. specific

heat c. joule f. heat

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*capacity _____ 1.
quantity of heat
needed to raise the
temperature of 1 g
of water by 1 °C
_____ 2. SI unit of
energy _____ 3.*

***AP Chemistry
Chapter 17 Outline -
SlideShare***

***In this lecture I'll
teach you how to
about the common***

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ion effect and how to perform pH calculations for common ion effect problems. I'll also teach you what buffered solutions are and how to ...

AP Chemistry Study Guide – Chapter 17, Electrochemistry
AP Chemistry
Chapter 17 Outline

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1. Chapter 17

***Additional Aspects
of Aqueous
Equilibria***

***Chemistry, The
Central Science ,
11th edition***

***Theodore L. Brown;
H. Eugene LeMay,
Jr.; and Bruce E.
Bursten John D.
Bookstaver St.***

***Charles Community
College Cottleville,***

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***A.P. Chemistry
Practice Test - Ch.
17: Electrochemistry
A ...***

***Chapter 17:
Electrochemistry. ...
You need to know
Redox reactions for
this chapter. Here is
a video that
summarizes redox:
YouTube Video.***

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***electrochemistry -
Selections from
Encyclopaedia
Britannica. A good
website that goes
into very detailed,
advanced (and
perhaps
unnecessary for the
AP Chemistry High
School Course) ...***

***Milstead, Millie / AP
Chemistry Acid-***

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Base Equilibria Ch

...

AP Chemistry.

APChem-Course

Syllabus 2016-2017.

APChem-Laboratory

Safety Code of

Conduct Contract. ...

AP Chem-CH14-HW-

Solutions. Chapter

15 – Chemical

Equilibrium. ...

Chapter 16 – Acid-

Base Equilibria. APC

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Solutions. Chapter
17 – Additional
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Equilibria. APChem-
CH17-HW-Solutions.
Chapter 20 –
Electrochemistry.***

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