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Chapter 19...Acids, Bases, and Salts ACID-BASE THEORIES Acids and bases are all around us and part of our everyday life (ex. bodily functions, vinegar, carbonated drinks, citrus fruits, car batteries, antacids, cleaning products, etc.). Acids and bases have the following properties: Acids Bases Tart or sour taste Bitter taste

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You will compare acids and bases and understand why their strengths

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vary. You will define pH and pOH and calculate the pH and pOH of aqueous solutions. You will calculate acid and base concentrations and determine concentrations experimentally. You will explain how buffers resist changes in pH.

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According to this Odel, in an acid-base reaction, each acid has a (17) , and each base has a (18) Study Guide for Content Mastery Chemistry: Matter and Change Chapter 19 . Name 19 Date Class Section Strengths of Acids and Bases In your textbook, read about strengths of acids.

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A B; hydrolyzing salts: compounds derived from the reaction of a strong base with a weak acid or a strong acid with a weak base or a solution of a weak base and one of its salts

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Chapter 19 “ Acids, Bases, and Salts ... Figure 19.22, page 615.

Neutralization. The solution of known concentration is called the standard solution. added by using a buret. Continue adding until the indicator . changes color.

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View Chemistry - Chapter 19 Acids, Bases, and Salts from SCIENCE 120 at Eastern Michigan University. Chapter 19 Acids, Bases, and Salts Chemistry Section 19.1 Acid-Base Theories OBJECTIVES: Define

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Acids produce what type of ions?  $H^+$  Acids taste \_\_\_\_ Acids turn \_\_\_\_ litmus paper \_\_\_\_.

1) sour 2) blue, red pH of an acid is \_\_\_\_ than 7.

Less What are two properties of strong acids? Good conductors and strong electrolytes Give an example of at least one strong acid. Any of the following: [...]

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Section 19.1 – Acid-Base Theories Acids have a sour taste, change the

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color of an indicator, can be strong or weak electrolytes in aqueous solution, and react with metals.

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