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- Notes

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Organisms are  
composed of  
matter. An  
element is a  
substance that  
cannot be broken

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Context Of Life  
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down into other  
substances by

chemical  
reactions. A  
compound is a  
substance  
consisting of  
two or more  
elements in a  
fixed ratio.

Table salt  
(sodium chloride  
or NaCl) is a

...

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of Life -  
BIOLOGY JUNCTION  
the elements or  
compounds  
produced by a  
chemical  
reaction  
Chemical  
Equilibrium In a  
chemical  
reaction, the

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state in which  
the rate of the  
forward reaction  
equals the rate  
of the reverse  
reaction, so  
that the  
relative  
concentrations  
of the reactants  
and products do  
not change with  
time

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Chemical Context

of Life\* - HCC  
Learning Web  
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of Life 19 12)  
Different atomic  
forms of an  
element contain  
the same number  
of protons but a  
different number  
of neutrons.

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Spiritual Realm  
- Duration:  
14:19. Russell  
Brand  
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Chapter 2: THE  
CHEMICAL CONTEXT  
OF LIFE

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of Life. A  
covalent bond  
between atoms  
that differ in e  
lectronegativity  
. The shared  
electrons are  
pulled closer to

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the more  
electronegative  
atom, making it  
slightly  
negative and the  
other atom  
slightly  
positive.

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Reading Guide

The Chemical  
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This chapter covers the basics that you may have learned in your chemistry class. Whether your teacher goes over this chapter, or assigns it for

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you to review on  
your

AP Bio Ch 02 -  
The Chemical  
Context of Life  
(Part 1)  
Chapter 2 The  
Chemical Context  
of Life This  
chapter presents  
basic chemical  
principles for  
understanding

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the chemical  
context of  
living  
organisms, from  
atomic structure  
to the nature of  
chemical bonds  
and an  
introduction to  
chemical  
equilibrium.  
These questions  
focus on  
elements most

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important to

life or the

study of life,

including the

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BIOLOGY I.

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Chemical Context  
of Life Life Has

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a Unique  
Chemistry •  
Chemists  
recognize 92  
elements  
occurring in  
nature. About 25  
are known to be  
essential to  
life, with 4  
predominant  
elements: oxygen  
(O), carbon (C),  
hydrogen (H),

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Context Of Life  
and nitrogen (N)

make up 96% of  
living matter. •  
Other important  
elements in  
living

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context of life

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• Atoms with  
more than 2  
electrons must  
have the extra

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electrons in  
higher shells. o

Lithium, with 3  
electrons, has 2  
electrons in the  
first shell and  
1 in the second  
shell. • The

second shell can  
hold up to 8  
electrons. o

Neon, with 10  
total electrons,  
has 2 in the

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first shell and  
8 in the second,  
thus filling  
both shells.

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Science ...

2. The organism  
responsible for  
this pruning is  
the ant species.  
This organisms.

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use formic acid  
as a herbicide.

3. This chapter  
focuses on the  
chemical  
components that  
make up all  
matter. I.

Matter consists  
of Chemical  
Elements in Pure  
Form and in  
Combination.  
Called

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Compounds. 1.  
Organisms are

composed of  
matter. 2.

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of Life This  
chapter covers  
the basics that

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learned in your  
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chapter, or  
assigns it for  
you to review on  
your own, the  
questions that  
follow should  
help you focus  
on the most

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important

points. Concept

2.1 Matter

consists of

chemical

elements in pure

form and in

combinations

called compounds

1.

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Chemical Context

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2) concentration  
of reactants 3)  
presence of  
chemical  
catalysts  
(enzymes) 4)  
temperature

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3. Elements and  
Compounds •

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Matter is made  
up of elements •

An element is a  
substance that  
cannot be broken  
down to other  
substances by  
chemical

reactions • A  
compound is a  
substance  
consisting of  
two or more  
elements in a

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fixed ratio • A  
compound has  
characteristics  
different...

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\*Lecture notes  
are to be used  
as a study guide

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only and do not  
represent the

comprehensive  
information you  
will need to  
know for the  
exams. A

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Connection to  
Biology There is  
relevance of  
chemistry to the  
study of life.

Organisms and

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their  
environments are  
natural systems  
to

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chapter covers  
the basics that

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questions that  
follow should  
help you focus  
on the most

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important  
points.

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of Life ...  
The chemical  
behavior of an  
atom is  
determined by  
the distribution  
of electrons in  
the atom's

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electron shells.  
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2.1 Matter  
consists of  
chemical  
elements in pure  
form and in  
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called compounds  
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