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Goals : To gain
an understanding
of : 1.

Stoichiometry.

2. Limiting
reagents and
percent yield.

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NOTES:

Stoichiometry is the calculation of chemical quantities from balanced equations. The four quantities involved in stoichiometric calculations are:

mc06se cFMsr i-

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SHORT ANSWER

Answer the
following
questions in the
space provided.

1. Given the
following
equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$?

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20(g) 4 a. What
is the value of
the coefficient
x in this

equation? 40.07

g/mol b. What is
the molar mass

of C₃H₄? 2 mol
O₂:1 mol H₂O

c. What is the
mole ratio ...

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Name _____ Class _____ Date _____

Chapter Test B,

continued 18.

When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is _____ than 7.

19. A is a highly _____

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