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NOTES - IB dead**

**Topic 1 -**

**Stoichiometry (HL  
and SL have the  
same material) SL  
Past Paper. Topic 2 -  
Atomic Theory. SL  
Past Paper. HL Past  
Paper. Topic 3 -  
Periodicity. SL Past  
Paper. HL Past  
Paper. Topic 4 -  
Bonding. SL Past**

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***Paper. HL Past  
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***Monday , May 11***

***2020. About Us; ...***

***0620 November 2012***

***Paper 51***

***Instructions. 0620***

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***Question Paper 51.***



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**Paper 52**

**Instructions.**

***Examination paper  
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pack - IB Documents  
The HL IB chemistry  
exam consists of  
three papers. Paper  
1: multiple choice - 1  
hour (20% - 40  
marks) There are 40  
questions and you***

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***are given 60 minutes  
to complete them.***

***This works out to an  
average of 1 minute  
and 30 seconds per  
question.***

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Official***

***Analyzing 7  
common multiple-  
choice questions***

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***often seen on the IB chemistry HL exam paper 1. The topics are oxidation and reduction, forming a buffer solution from different concentrations of weak ...***

***Question 1 — IB Chemistry HL — May 2017 TZ1 Paper 1 — Past IB Exams***

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IB HL topics have  
been folded into SL  
topics, so topic 2  
and 12 are both in***

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***HL topic 2 revision  
booklets NB SL***

***Topic 2 has been  
expanded to include  
more content that  
was previously  
examined only in  
HL, additional  
revision booklets  
have been included  
below to give you  
practice with these  
HL questions that  
are now part of SL***

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**IB Documents -  
Resources  
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***In paper 1 there are  
30 (at SL) or 40 (at  
HL) multiple-choice  
questions. Paper 2  
contains short-  
answer and  
extended-response  
questions on the  
core (and Additional  
Higher Level (AHL)***

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*material at HL).*

***Paper 3 has two sections; Section A contains one data-based question and several short-answer questions on experimental work on the core (and AHL material at HL).***

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**QUANTITATIVE**

**CHEMISTRY. TOPIC**

**2: ATOMIC**

**STRUCTURE. TOPIC**

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**3: PERIODICITY.**

**TOPIC 4: BONDING.**

**TOPIC 5:**

**ENERGETICS.**

**TOPIC 6: KINETICS.**

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**Question 5 — IB  
Chemistry HL — May  
2017 TZ1 Paper 1 —  
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***SL May 2018 TZ2.***

***Paper 1. Paper 1.***

***Paper 2. Chemistry***

***SL May 2018 TZ1.***

***Paper 2. Paper 1.***

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**Paper 3. Chemistry**

**SL May 2017 TZ2. ...**

**Chemistry HL May**

**2016 TZ0 ...**

**Chemistry 2012**

**Paper 3 HL**

**[3] F1. Crude oil**

**contains many**

**hydrocarbons,**

**including hexane,**

**CH<sub>6</sub>14. (a) Outline**

**how crude oil was**



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**formed..... [3] (b) The equation for the complete combustion of hexane is shown below.  $C_6H_{14}(g) + 9\frac{1}{2}O_2(g) \rightarrow 6CO_2(g) + 7H_2O(g)$  Determine a value for the enthalpy of combustion of hexane using the following enthalpy of formation data ...**

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**OpB 2 Chem OpB 3**

**Chem OpB 4 Chem**

**OpB 5 Chem OpB 6**

**Chem OpB 7 Chem**

**OpB 8 Chem OpB 9**

**Chem OpB 10 Chem**

**NoS HL 0.0 13.5 12.0**

**9.0 1.5 5.3 11.3 3.8**

**11.3 1.5 30.8**

**Name IB ALL HL EQ**

**P3 Nature of**

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B questions**

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