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Science is a wonderful subject to learn about – whether it's for career purposes or you're merely curious about the ins and outs of our universe, including what makes up the substances that comprise who we are and what we see daily. Today we'll be looking at science from a chemical standpoint, seeing if you can identify basic chemistry concepts and balance equations. How will you perform ...

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Test 1) Answer the following questions in the spaces provided on the extreme right of the page. You do not have to show your work.

Significant figures and appropriate units are required. /6 a) 1.0 g of NaCl contains _____ grams of sodium b) The mass of 2.4×10^{25} atoms of sodium is: c) 5.0 g of CaCO_3 contains _____ moles of ...

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test for radioactivity. d. use of X rays. ____ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c. ... Chemistry Chapter 5 Exam ...

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AP Chemistry: Practice Test, Ch. 5. - Gases Name____ MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1)According to kinetic-molecular theory, in which of the following gases will the root-mean-square speed of the molecules be the highest at 200eC? A)SF6 B)H2O C)HCl D)Cl2

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