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e. molar mass of $\text{Fe}(\text{NO}_3)_3 = 241.9 \text{ g}$ 272 g $\text{Fe}(\text{NO}_3)_3$ 1 mol 241.9 g = 1.12 mol $\text{Fe}(\text{NO}_3)_3$ f. molar mass of $\text{FeSO}_4 = 151.9 \text{ g}$ 12.7 mg FeSO_4 1 g 1000 mg 1 mol 151.9 g = 8.36 10^{-5} mol FeSO_4 4 g. molar mass of $\text{LiOH} = 23.95 \text{ g}$ 6.91 103 g LiOH 1 mol 23.95 g = 288.5 = 289 mol LiOH 14. a. molar mass of $\text{CaCO}_3 = 100.1 \text{ g}$ 2.21 10^{-4} mol CaCO_3 100.1 g 1 mol = 0 ...

Chemistry Chapter 9 Study Guide Answer Key
Chapter 9 - Chemical Calculations and Chemical Formulas 119 Chapter 9

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Matching Match each item with the correct statement below. Match each item with the correct statement below. a. monatomic ion f. cation b. acid g. binary compound c. base h. anion d. law of definite proportions i. polyatomic ion e. law of multiple proportions ____ 1.

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Chapter 9 Chemical C and C Formulas - Mark Bishop
CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88%
The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed with 12.0 mol of H

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"Inner energy levels tend to block the attraction of the nucleus for the valence electrons; reason why electronegativity goes down in value in a group/family."- is called the _____.

NCERT Solutions for Class 11 Chemistry Chapter 9 Hydrogen
Answers may include release or absorption of energy, change in color, change in odor, formation of a gas, or formation of a solid. 9.

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Chemistry \ Chemistry Chapter 9 Answers. Chemistry Chapter 9 Answers Flashcard. Flashcard maker : shippo. In which pair do both compounds exhibit predominantly ionic bonding? RbCl and CaO. The following representation of an atom is called. A Lewis dot structure.

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