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Modern Chemistry CHAPTER 22
HOMEWORK 22-1

CHAPTER 7 • HOMEWORK 7-1

VOCABULARY 1. false 2. true 3. false 4.
false 5. true SKILL BUILDER AgCl —In
the compound, there is one atom of silver
for each atom of chlorine. Ba(OH)_2
—The subscript 2 refers to everything

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inside the parentheses, so there are two hydroxide ions for

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Modern Chemistry • CHAPTER 8
HOMEWORK 8-3 (pp. 248–250)

VOCABULARY Complete each sentence.

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1. A chemical equation is like an algebraic equation because they both _____. 2. The coefficients of a chemical reaction indicate _____. 3.

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HOMEWORK 2-1

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Modern Chemistry • CHAPTER 2

HOMEWORK 2-2 (pp. 33–36)

VOCABULARY Write true or false. 1. SI units are part of a system of measurement used only in North America. _____ 2. A quantity, such as volume, is not the same thing as a

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measurement. _____ 3. The SI base unit for amount of substance is kilogram. _____ 4.

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person is known as the father of modern
chemistry. The science that studies all
living systems is known as.

HOMEWORK 7-2

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2 answers. Moral: As a mere proletariat,
his ignorance is a temptation to the
stronger race; as a voter, it is a danger to

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them which it becomes their interest to remove.

Modern Chemistry Chapter 2
Measurements and Calculations ...
Modern Chemistry • CHAPTER 22
HOMEWORK 22-2 (pp. 702–704)
VOCABULARY Write true or false. 1.

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The band of stability represents the stable nuclei cluster over a range of electron-proton ratios. _____ 2. The stability of a nucleus is greatest when the nucleons are in a 1:1 ratio. _____ 3. 2, 20, 50, and 126 are all magic numbers. _____ 4.

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HOMEWORK 8-1

2. Which statement is true? a. In a number with no zeros, all the digits are significant. b. In chemistry, the term significant means "certain." c. Both significant and insignificant digits are always reported in the results of an

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experiment. d. Zeros at the end of a number are not significant digits.

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