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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the

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Modern Chemistry Chapter 2 Section 3 Review Answers
Holt Chemistry 40 Matter and Energy Section: Measurements and Calculations in Chemistry In the space provided, write the letter of the term or phrase that best answers the question. ____ 1. A measurement is said to have good precision if it a. agrees closely with an accepted standard. b. agrees closely with other measurements of the same quantity.

Kristy Blankenship - Boyd County High School (9-12)
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Assessment Chapter Test A
Holt McDougal Modern Chemistry 2 Chapter Test Chapter Test A, continued ____ 3. The periodic law states that a. no two electrons with the same spin can be found in the same place in an atom. b. the physical and chemical properties of the elements are functions of their atomic number. c. wave patterns repeat at regular intervals.

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Assessment Chapter Test A
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Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued ____13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. ____14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

6 Chemical Bonding
A substance, made up of two or more atoms that are chemically bonded, that can be broken down into simpler, stable substances

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