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**701 Name: Kelly R. Lab**

**results: 1 doz grains of rice =  
0.3 g (Use this fact as a  
conversion factor.)? grains of  
rice = 1.94 g Avogadro's**

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**Number - the mole = the  
number  $6.022 \times 10^{23}$  Molar  
Mass the mass of one mole of  
any pure substance molar  
mass of an element (g/mol)  
equals the atomic mass of the  
element in grams ...**



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**Day 12 3, 1, 5, 2, 4 B B B, A,  
A, A, B B, A, B, A C D B**

**Experiment is to determine whether the hypothesis is correct. If not, alter the hypothesis and conduct additional experiments to test the new hypothesis. (It is**

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**often an ongoing process for  
scientists.) Or similar answer.  
Safe,...**

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**carbon are called \_\_\_\_\_.  $C_nH_{2n+2}$  is the general form for the \_\_\_\_ series of hydrocarbons. The names of this series ...**

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