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your answer to
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enthalpy of fusion
given in 'Model 2 for
each substance is a
measure of the
energev (in k)) needed
to melt a mole of that
substance. a. Which
typc of solid has the
10N\Qst enthalpies of
fusion.' b.

Chem 116 POGIL

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Worksheet - Week 4
Properties of
Solutions

All of the answers can be found online for almost every single POGIL..... Currently, there is a significant amount of discussion on teaching list serves about the frustration of people posting answer keys online and students

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checking the internet instead of doing the work. Many teachers, myself included, experience this frustration.

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Worksheet - Week 5

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Properties of
Solutions Why? We
have discussed the
properties of gases,
liquids, and solids as
pure substances. But
most of our use of
these phases is in the
form of mixtures.
Solutions, which are
homogeneous
mixtures, ... □ Be able
to predict solubility,
based on the

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Pool Answers

intermolecular forces
of the components

Common Ion Effect
on Solubility

4 POGIL Activities
for High School

Chemistry Extension
Questions Model 2

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90 80 70 60 50 40 30

20 10 0 0 10 20 30

Temperature ($^{\circ}\text{C}$)

Solubility (g

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solute/100 g water) 40
50 60 Substance A
Substance B 12.
According to the
graph in Model 2,
what is the solubility
of Substance A at 30
oC? 13.

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to Chemistry □ Safety
First ... Solubility and
Solutions □ Saturated
and Unsaturated
Solutions □ Solubility □
Molarity
Thermochemistry ...
In your answer,
consider the external
and internal pressure
data given in Model 1.

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units of solubility on
solubility curves from
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4 □ Activities for AP*

Chemistry POGIL 14.

The solubility product constant, K_{sp} , for silver carbonate, Ag_2CO_3 , is 8.10×10^{-12} . Compare the solubility of silver carbonate in water to its solubility in a 0.100 M solution of silver nitrate.

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comparison with
calculations similar to
those in Model 1.

2 POGIL Saturated
and Unsaturated
Solutions and
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12 - Water and
Aqueous Solutions -
CSISD Chemistry

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Chem 115 POGIL

Worksheet - Week 5

Solubility and Solution

Reactions Why? So

far, we have

considered

stoichiometric

relationships on the

basis of masses and

moles. Often it is

more convenient to

add a substance into

a reaction mixture in

the form of a solution,

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rather than as a weighed sample of material. We can still make predictions of ...

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Dimensional Analysis

(Calculations in

Chemistry) 3 - Atomic

Structure. 4 - The

Periodic Table. 5- The

Mole (a central unit in

Chemistry) ... 3

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