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calculation of absolute entropies of the substances. Mathematically, LimT ...

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Heat Engines, Entropy, and the Second Law of Thermodynamics One mole of an ideal gas is heated at constant pressure from 0oC to 200oC. (a) Calculate work Page 9/16

done. (b) If the gas were expanded isothermally & reversibly at 0°C from 1 atm to some other pressure Pt, what must be the final pressure if the maximum ...

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Page 10/16

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Chapter 17. Work, Heat, and the First Law of Thermodynamics
Thermodynamics 22.2 Heat Pumps and Refrigerators 22.3 Reversible and Irreversible Processes 22.4 The Carnot Engine 22.5 Gasoline and Diesel Engines 22.6 Entropy 22.7 Entropy Changes in Irreversible Processes Scale 22.8 Entropy on a Microscopic Heat Engines, Entropy, and the Second Law of Thermodynamics ANSWERS TO QUESTIONS

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The First Law of Thermodynamics Work and heat are two ways of transfering energy between a system and the environment, causing the system's energy to change. If the system as a whole is at rest, so that the bulk mechanical energy due to translational or rotational motion is zero, then the

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